

Exam #1 Key

① weight %: Assume 1 g/mL, so 250 mL = 250 g
$$\Rightarrow \frac{45.1 \times 10^{-6} \text{ g}}{250. \text{ g}} \times 100 = 1.80 \times 10^{-5} \%$$

ppm: 1 ppm = 1 $\mu\text{g/mL}$
$$\Rightarrow \frac{45.1 \mu\text{g}}{250. \text{ mL}} = 0.180 \text{ ppm}$$

ppb: 1 ppb = 1 $\mu\text{g/L}$
$$\Rightarrow \frac{45.1 \mu\text{g}}{.250 \text{ L}} = 180. \text{ ppb} = 1.80 \times 10^2 \text{ ppb}$$

② $\mu = 77.5$ $\bar{x} = 76.73$ $t = 2.262$
 $S = 2.284$

$$\bar{x} \pm \frac{tS}{\sqrt{n}} = 76.73 \pm \frac{2.262(2.284)}{\sqrt{10}} = 76.73 \pm 1.63$$

$$\Rightarrow 95\% \text{ C.I.: } 76.7 \pm 1.6 = \underbrace{75.1 \text{ to } 78.3}_{\text{contains } \mu}$$

\Rightarrow results do agree at the 95% C.I.

② (another way)

$$\mu = \bar{x} \pm \frac{tS}{\sqrt{n}}$$

$$t_{\text{calc}} = \frac{|\mu - \bar{x}| \sqrt{n}}{S} = \frac{|77.5 - 76.73| \sqrt{10}}{2.284}$$

$$t_{\text{calc}} = 1.07$$

$$t_{\text{tab}} = 2.262$$

$t_{\text{calc}} < t_{\text{tab}} \Rightarrow$ no significant difference

$$\textcircled{3} \quad \frac{([23.5811_g (\pm 0.0003)] - [22.1559_g (\pm 0.0003)]) \left(\frac{1 \text{ mol}}{121.34_g (\pm 0.01)}\right)}{[100.0 \text{ mL} (\pm .5)] \left(\frac{1 \text{ L}}{1000 \text{ mL}}\right)}$$

$$= 0.117456 \text{ M} \pm ?$$

$$23.5811 - 22.1559 = 1.4252 \text{ g}$$

$$e = \sqrt{(0.0003)^2 + (0.0003)^2} = .000424$$

$$\frac{[1.4252 \text{ g} (\pm 0.000424)] \left(\frac{1}{121.34 (\pm 0.01)}\right)}{[100.0 \pm .5] \left(\frac{1}{1000}\right)}$$

$$\frac{e}{.117456} = \sqrt{\left(\frac{.000424}{1.4252}\right)^2 + \left(\frac{.01}{121.34}\right)^2 + \left(\frac{.5}{100}\right)^2}$$

$$e = .000588$$

$$\Rightarrow \boxed{0.1174_6 (\pm 0.0005_9) \text{ M}}$$

$$\textcircled{4} \quad \% \text{ Fe in compound} = \frac{55.85}{392.14} \times 100 = 14.24_2 \% \text{ Fe}$$

$$.1424_2 (.0729 \text{ g}) = 0.01038_3 \text{ g Fe} = 10.38_3 \text{ mg Fe}$$

$$\frac{10.38_3 \text{ mg}}{1 \text{ L}} = 10.38_3 \text{ ppm}$$

$$M_1 V_1 = M_2 V_2$$

$$(10.38_3)(10) = M_2(50)$$

$$\boxed{M_2 = 2.08 \text{ ppm}}$$

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$$t_{\text{calc}} = \frac{|\bar{x}_1 - \bar{x}_2|}{s_p} \sqrt{\frac{n_1 n_2}{n_1 + n_2}}$$

$$s_p = \sqrt{\frac{s_1^2(n_1 - 1) + s_2^2(n_2 - 1)}{n_1 + n_2 - 2}}$$

$$s_p = \sqrt{\frac{.093^2(6-1) + .087^2(5-1)}{6+5-2}} = .09038$$

$$t_{\text{calc}} = \frac{|8.63 - 8.84|}{.09038} \sqrt{\frac{6 \cdot 5}{6+5}} = \boxed{3.84}$$

$$t_{\text{tab}} = 2.262 \text{ (9 d.f., 95\%)}$$

$t_{\text{calc}} > t_{\text{tab}} \Rightarrow$ difference is significant.

⑥ a) Volumetric unknown: $\bar{x} = 61.04$ $n = 46$ $\mu = 59.85$
 $S = 1.89$ $t_{95\%} = 2.014$ $t_{99\%} = 2.690$

$$t_{\text{calc}} = \frac{|\mu - \bar{x}|}{S} \sqrt{n}$$

$$= \frac{|59.85 - 61.04|}{1.89} \sqrt{46} = 4.27$$

$t_{\text{calc}} > t_{\text{tab}}$ for both 95% + 99% CI
 \Rightarrow method does not give correct result

Gravimetric: $\bar{x} = 60.33$ $n = 46$ $\mu = 59.85$
 $S = 0.56$ $t_{95\%} = 2.014$ $t_{99\%} = 2.690$

$$t_{\text{calc}} = \frac{|\mu - \bar{x}|}{S} \sqrt{n} = \frac{|59.85 - 60.33|}{0.56} \sqrt{46} = 5.81$$

$t_{\text{calc}} > t_{\text{tab}}$ for both 95% + 99% CI
 \Rightarrow method does not give the correct result

Alternatively: $CI = \bar{x} \pm \frac{t_s}{\sqrt{n}}$

Volumetric: $\frac{t_s}{\sqrt{n}} = 0.561$ for 95%
 \Rightarrow ~~60.47~~ 60.48 to 61.60
 $\frac{t_s}{\sqrt{n}} = 0.749$ for 99%
 \Rightarrow 60.29 to 61.79
 } neither CI contains μ

Gravimetric: $\frac{t_s}{\sqrt{n}} = 0.166$ for 95%
 \Rightarrow 60.16 to 60.50
 $\frac{t_s}{\sqrt{n}} = 0.222$ for 99%
 \Rightarrow 60.11 to 60.55
 } neither CI contains μ

b) F: test

$$F_{\text{calc}} = \frac{1.89^2}{0.56^2} = 11.4 \quad F_{\text{tab}} = 1.64$$

\Rightarrow variances are different at the 95% C.L.

c) T-test assuming unequal variances:

$$\left. \begin{array}{l} t_{\text{calc}} = 2.43 \\ t_{\text{tab}} = 2.01 \end{array} \right\} t_{\text{calc}} > t_{\text{tab}} \Rightarrow \text{results are different at the 95\% C.L.}$$

d) Volumetric: Standard: $0.10034 (\pm .0015) \text{ mol/L}$
unknown: $0.03360 (\pm .000964) \text{ g/mL}$

Calculation:

$$\left(\frac{0.10034 \text{ mol NaOH}}{\text{L NaOH}} \right) \left(\frac{1 \text{ L NaOH}}{1000 \text{ mL NaOH}} \right) \left(\frac{1 \text{ mL NaOH}}{0.03360 \text{ g solid}} \right) \left(\frac{1 \text{ mol KHP}}{1 \text{ mol NaOH}} \right) \left(\frac{204.221 \text{ g KHP}}{1 \text{ mol KHP}} \right) = 0.60987 \text{ g KHP/g solid}$$

Error: $\frac{e}{0.60987} = \sqrt{\left(\frac{0.0015}{0.10034} \right)^2 + \left(\frac{0.000964}{0.03360} \right)^2} \Rightarrow e = 0.0197$
 $\Rightarrow 0.6099 (\pm 0.0197) = \boxed{61.0 (\pm 2.0)\%}$

Gravimetric: Standard: $0.10002 (\pm .00120) \text{ mol/kg soln}$
unknown: $0.033862 (\pm .000310) \text{ g unk/g NaOH soln}$

Calculation:

$$\left(\frac{0.10002 \text{ mol NaOH}}{\text{kg NaOH soln}} \right) \left(\frac{1 \text{ kg NaOH soln}}{1000 \text{ g NaOH soln}} \right) \left(\frac{1 \text{ g NaOH soln}}{0.033862 \text{ g unk}} \right) \left(\frac{1 \text{ mol KHP}}{1 \text{ mol NaOH}} \right) \left(\frac{204.221 \text{ g KHP}}{1 \text{ mol KHP}} \right) = 0.60322 \text{ g KHP/g unknown}$$

Error: $\frac{e}{0.60322} = \sqrt{\left(\frac{0.00120}{0.10002} \right)^2 + \left(\frac{0.000310}{0.033862} \right)^2} \Rightarrow e = 0.00910$
 $\Rightarrow 0.6032 (\pm 0.00910) = \cancel{60.32\%} = \boxed{60.32 (\pm 0.9)\%}$

7)

Patient	Method A	Method B
1	1044	1028
2	720	711
3	845	820
4	800	795
5	957	935
6	650	639

t-Test: Paired Two Sample for Means

	Method A	Method B
Mean	836	821.33333
Variance	21466.8	20349.067
Observations	6	6
Pearson Correlation	0.9989157	
Hypothesized Mean Difference	0	
df	5	
t(calc) = t Stat	4.6277348	
P(T<=t) one-tail	0.0028477	
t Critical one-tail	1.475884	
P(T<=t) two-tail	0.0056954	
t(tab) = t Critical two-tail	2.0150484	

t(calc) > t(tab), so there is a significant difference

OR (alternatively):

Patient	Method A	Method B	di	$(d_i - d_{i,ave})^2$
1	1044	1028	16	1.78
2	720	711	9	32.11
3	845	820	25	106.78
4	800	795	5	93.44
5	957	935	22	53.78
6	650	639	11	13.44

$$d_{i,ave} = 14.7$$

$$\Sigma[(d_i - d_{i,ave})^2] = 301.33$$

$$n = 6$$

$$s_d = 7.76$$

$$t_{calc} = 4.63$$

$$t_{tab} = 2.015$$

8)

Protein (µg)	Abs (595 nm)	$(x_i - x_{ave})^2$
0.000	0.0021	25.23
2.024	0.1517	8.99
4.018	0.2947	1.01
6.135	0.4342	1.24
7.975	0.5701	8.72
9.984	0.7045	24.61
Sum =		69.80

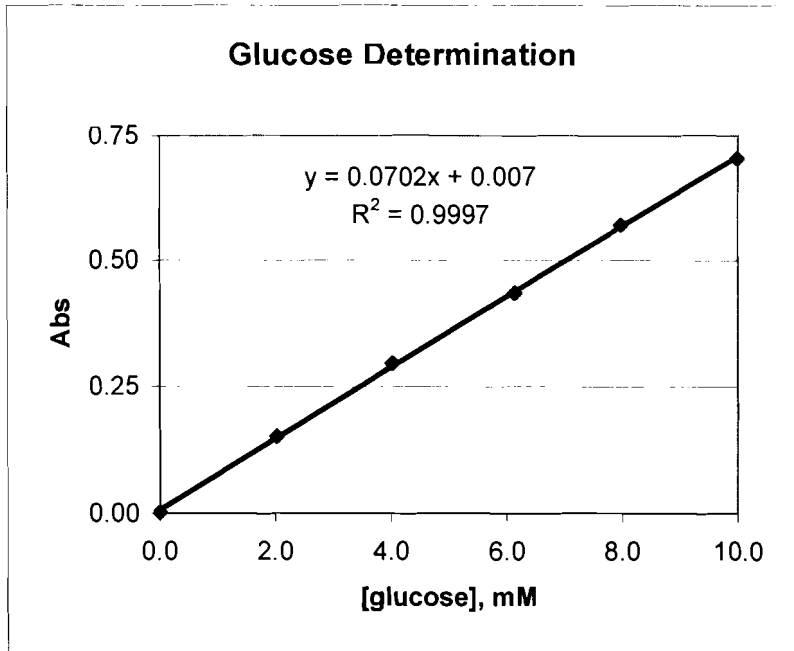
SUMMARY OUTPUT

Regression Statistics	
Multiple R	0.999859474
R Square	0.999718968
Adjusted R Square	0.99964871
Standard Error	0.004916688
Observations	6

ANOVA

	df	SS	MS	F	Significance F
Regression	1	0.34397538	0.34397538	14229.25006	2.96199E-08
Residual	4	9.66953E-05	2.41738E-05		
Total	5	0.344072075			

	Coefficients	Standard Error	t Stat	P-value	Lower 95%	Upper 95%	Lower 95.0%	Upper 95.0%
Intercept	0.006951958	0.003572993	1.94569571	0.123566294	-0.002968262	0.016872178	-0.00296826	0.016872178
Protein (mg)	0.070201362	0.000588511	119.2864203	2.96199E-08	0.068567394	0.071835331	0.068567394	0.071835331



$s_y = 0.004916688$
 $|m| = 0.070201362$
 $b = 0.006951958$
 $k = 3$
 $n = 6$

$y(ave) = 0.3596$
 $y = 0.350$
 $x(ave) = 5.02$

$x = 4.887$ $E_{30} = (C38 - C34) / C38$
 $s_x = 0.0495$
 $(C32 / C33) * \text{SQRT}((1 / C35) + (1 / C36) + ((C38 - C37)^2) / ((C33^2) * \text{SUM}(D4 : D9)))$

4.88₇ (± 0.05₀) mM